

Chemical Oceanography MARS 8020, Spring 2017

Tue/Thu 2 – 3:15 pm

Room 261, Marine Sciences Bldg

Instructors: Christof Meile (cmeile@uga.edu); Patricia Medeiros (medeiros@uga.edu)

Content: This course will cover most of the fields of chemical oceanography/marine chemistry. Students are expected to have had knowledge in general chemistry and oceanography, and high motivation in learning. There will be some extra reading materials, reasonable amount of home works, and two tests and a final exam.

Grades: Two tests and a Final Exam, 25% each; home work 20%, and paper discussion & classroom participation 5%.

Recommended Textbook: Emerson S. & Hedges J. 2008. Chemical Oceanography and the Marine Carbon Cycle. Cambridge University Press. ISBN: ISBN: 9780521833134

Academic Honesty: As a University of Georgia student, you have agreed to abide by the University's academic honesty policy, "A Culture of Honesty," and the Student Honor Code. All academic work must meet the standards described in "A Culture of Honesty" found at: <https://ovpi.uga.edu/academic-honesty/academic-honesty-policy>. Lack of knowledge of the academic honesty policy is not a reasonable explanation for a violation. Questions related to course assignments and the academic honesty policy should be directed to the instructor.

Plagiarism ("to take ideas, writings, etc. from another and pass them off as one's own", Webster's New World Dictionary) will not be tolerated. There are several forms of plagiarism, ranging from outsourcing your work to somebody else, to slight rewording of a published text or summarizing a text without citing it. If you are in doubt consult with the instructor before you hand something in.

Changes to the Course Syllabus: The course syllabus is a general plan for the course; deviations announced to the class by the instructor may be necessary. Failure to regularly attend class may result in your being uninformed about changes in the course content or timing of assignments. Students who miss class are responsible for all announcements and assignments given in lecture.

Access Statement: The University of Georgia is committed to providing access for all people with disabilities and will provide accommodations if notified prior to the start of the semester. Please contact the Disability Resource Center if you will need a sign language interpreter, assisted listening device, or other classroom accommodations. If you would like to discuss classroom and/or testing accommodations, please discuss your needs with the instructors of record as soon as possible.

Date	Topic	Instructor
	<i>Water & Ionic Interactions</i>	
5-Jan	About the course 1. structure of water	CM
10-Jan	2. ions in water, ionic strength 3. ion-ion interaction 4. inorganic speciation in seawater	CM
	<i>Chemical Speciation</i>	
12-Jan	1. complexation chemistry & chemical constants, solubility 2. processes controlling trace metal distribution (I); trace metal (II) organics/siderophores	CM
17-Jan	3. surface speciation, sorption etc.	CM
	<i>Redox Reactions</i>	
19-Jan	1. pe, Eh, and the Nernst equation 2. understanding pe-pH diagrams	CM
24-Jan	3. Balancing equations 4. role of O ₂ in marine redox, real world demonstrations of redox in the ocean	CM
	<i>Dissolved Gases</i>	
26-Jan	5. gas solubility and rates of gas exchange across the air-sea interface	CM
31-Jan	review session/buffer	CM
2-Feb	First Exam	CM
	<i>Major Elements in the Oceans</i>	
7-Feb	1. major composition of seawater 2. weathering reaction and river fluxes	CM
	<i>The Ocean CO₂ System</i>	
9-Feb	1. distributions and control mechanism of pH, pCO ₂ , Alkalinity and total CO ₂ 2. ocean acidification	CM
14-Feb	oceanic box models	CM
	<i>Productivity and Nutrient Biogeochemistry</i>	
16-Feb	1. photosynthesis, respiration, photic zone, sinking particles, etc.	CM
21-Feb	2. O ₂ and CO ₂ distribution in the oceans 3. O ₂ as a measure of biological production and respiration 4. the Redfield Ratio	CM
23-Feb	5. distributions of N, P, and Si in the world ocean	CM

28-Feb	6. Iron and other trace metal cycles	CM
<i>The Seafloor & longer timescales</i>		
2-Mar	1. glacial/interglacial differences in elemental cycles	CM
6-10-Mar	Spring break - No classes	
14-Mar	2. sediments: early diagenesis	CM
16-Mar	3. hydrothermal systems: vent chemistry	CM
21-Mar	Second Exam	CM
<i>Marine Organic Matter</i>		
23-Mar	1. Organic matter chemistry and cutoffs	PM
	2. Organic matter cycle	
28-Mar	3. Terrestrial vs microbial biomarkers	PM
	4. Organic matter fluxes and transport	
30-Mar	5. Organic matter degradation	PM
	6. Molecular record: OM preservation	
<i>Isotopes as Tracers</i>		
4-Apr	1. Stable isotopes: fractionation principles	PM
	2. Oxygen, carbon and nitrogen applications	
6-Apr	3. Radioisotopes principles	PM
	4. Radiocarbon applications - OM age	
11-Apr	5. Multi-tracers in marine organic chemistry studies	PM
<i>Processes at the Boundaries</i>		
13-Apr	1. Anthropogenic chemicals entering the marine environment I	PM
18-Apr	2. Anthropogenic chemicals entering the marine environment II	PM
20-Apr	3. Using anthropogenic organic compounds as markers	PM
25-Apr	Review Session - Organic	PM
28-April – 4-May	Third Exam (TBD)	PM

Important dates:

5-11 -Jan	Drop/Add
16-Jan	MLK Day
20-Mar	Withdrawal Deadline

Textbook

Emerson S. & Hedges J. 2008. Chemical Oceanography and the Marine Carbon Cycle. Cambridge University Press. ISBN: ISBN: 9780521833134

Other relevant books:

Pilson, M.E.Q. 2013. An introduction to the chemistry of the sea. 2nd edition. Cambridge University Press. ISBN: 9780521887076

Sarmiento J.L & Gruber, N. 2006. Ocean Biogeochemical Dynamics. Princeton University Press. ISBN: 9780691017075

Stumm W. & Morgan J.J. 1995. Aquatic Chemistry. 3d edition Wiley. ISBN: 978-0-471-51185-4

Morel F.M.M. & Hering J.G. 1993. Principles and Applications of Aquatic Chemistry. Wiley. ISBN: 978-0-471-54896-6

Broecker W. and Peng T.-H. 1982. Tracers in the Sea.

http://eps.mcgill.ca/~egalbrai/Earth_System_Dynamics/Tracers_in_the_Sea.html;

http://www.ldeo.columbia.edu/~broecker/Home_files/TracersInTheSea_searchable.pdf